## Reactions of 2-hydroxynaphthazarins with cyclohexane-1,2-dione as a new method for the synthesis of pyranonaphthazarin systems

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3,4-Dihydro-2*H*-naphtho[2,3-*b*]pyran-5,10-diones are produced by many micro- and macroorganisms in the living nature. <sup>1,2</sup> Most often, they are derivatives of parent 1,4-naphthoquinone or 5-hydroxy-1,4-naphthoquinone (juglone) and more rarely, of 5,8-dihydroxy-1,4-naphthoquinone (naphthazarin). Only a few examples of natural pyranonaphthazarins are known. <sup>1,2</sup> Approaches to their synthesis<sup>3,4</sup> are scanty and the first of them<sup>3</sup> is of low utility due to complexity and a large number of steps involved.

We found that reactions of 2-hydroxynaphthazarins (1) with cyclohexane-1,2-dione (2) open up a facile and a rather efficient route to pyranonaphthazarin systems.

## Scheme 1

За-с

**1, 3:** R = Cl (**a**), Me (**b**), H (**c**) **2:** R<sup>1</sup> = H, Me, (CH<sub>2</sub>)<sub>2</sub>OMe

The reactions of compounds **1a—c** with dione **2** under conditions of acid catalysis take place only in an alcohol solution; the best choice is heating in a solution in ethylene glycol or its monoethers. Irrespective of the group R<sup>1</sup> in R<sup>1</sup>OCH<sub>2</sub>CH<sub>2</sub>OH, the reaction gives pentacyclic pyranonaphthazarins **3a—c** in 40—85% yields (Scheme 1).

The structure of these products was established using various 1D (DEPT-135) and 2D (COSY-45, HMBC, HMQC)  $^{1}$ H and  $^{13}$ C NMR techniques. The relative stereochemistry of stereogenic centers, C(2), C(3), and C(4), in compounds **3a**—**c** was established by NOE experiments for **3a**, in particular, irradiation of the  $\alpha$ -axial proton at C(3) induces NOE effects for the signals of the  $\alpha$ -axial protons at C(15) (~7%), C(17) (~7%), and C(13) (~5%) and the  $\alpha$ -equatorial proton at C(4) (~16%), while irradiation of the  $\alpha$ -equatorial proton at C(4) entails NOE effects for the signals of  $\alpha$ -axial protons at C(3) (~9%) and C(15) (~4%). Study of the mechanism of this reaction is the subject of further research.

 $^{1}\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR spectra were recorded on a Bruker AVANCE DPX-300 spectrometer (300.13 MHz for  $^{1}\mathrm{H}$  and 75.47 MHz for  $^{13}\mathrm{C}$ ) in CDCl<sub>3</sub> (Me<sub>4</sub>Si as the internal standard). IR spectra were measured on a Bruker Vector 22 FT spectrometer in CHCl<sub>3</sub>. Mass spectra (EI, 70 eV) were run on an LKB-9000S instrument. The reactions were monitored and the purity of the obtained compounds was checked by TLC on Merck 60 F-254 plates in a hexane—Me<sub>2</sub>CO system (2:1). The individual compounds were isolated by preparative TLC (PTLC) on  $20\times20$  cm plates with an unbound silica gel layer (5–40  $\mu m$ , pretreated with oxalic acid) in a 2:1 hexane—Me<sub>2</sub>CO system. Melting points were determined on a Boetius hot stage and not corrected. Compounds 1b,c were prepared by analogy<sup>5</sup> with the synthesis of 1a. Commercial cyclohexane-1,2-dione (Aldrich) was used in the reactions.

Synthesis of pyranonaphthazarins 3a—c (general procedure). A solution of the required 2-hydroxynaphthazarin 1 (1.0 mmol), 1,2-diketone 2 (5.0 mmol), and anhydrous TsOH (0.05 mmol) in 5 mL of anhydrous 2-methoxyethanol was refluxed for 10—12 h. The reaction mixture was cooled, diluted with water (20 mL), and extracted with EtOAc (4×10 mL). The combined extracts were washed with saturated brine (3×20 mL) and dried with Na<sub>2</sub>SO<sub>4</sub> and the solvent was evaporated. The products were isolated by PTLC.

(2*S*,3*R*,4*R*)-7,8-Dichloro-2,3-ethylenedioxy-6,9-dihydroxy-2,4-trimethylene-3,4-dihydronaphtho[2,3-*b*]pyran-5,10(2*H*)-dione (3a). Yield 79%,  $R_{\rm f}$  0.54, m.p. 233—235 °C. Found (%): C, 52.58; H, 3.45; Cl, 17.51.  $C_{18}H_{14}Cl_2O_7$ . Calculated (%): C, 52.32; H, 3.42; Cl, 17.16. IR, v/cm<sup>-1</sup>: 3500—2100 (OH); 1611 (C=O, C=C); 1592 (C=C); 1560 (C=C); 1451; 1408; 1397; 1290; 1125.  $^{1}H$  NMR, &: 1.39—1.97 (m, 5 H, C(16)H<sub>ax</sub>, C(16)H<sub>eq</sub>, C(15)H<sub>ax</sub>, C(17)H<sub>eq</sub>); 2.18 (ddt, 1 H,

 $C(15)H_{eq}$ , J = 1.3 Hz, J = 1.3 Hz, J = 5.3 Hz, J = 13.7 Hz); 3.54 (q, 1 H, C(4)H, J = 3.4 Hz); 3.66 (dd, 1 H, C(12)H<sub>eq</sub>, J =2.7 Hz, J = 12.0 Hz); 3.77 (dd, 1 H, C(13)H<sub>eq</sub>, J = 3.2 Hz, J =12.0 Hz); 3.78 (d, 1 H, C(3)H, J = 3.4 Hz); 3.91 (ddd, 1 H,  $C(13)H_{ax}$ , J = 2.7 Hz, J = 12.0 Hz, J = 12.0 Hz); 4.21 (ddd, 1 H,  $C(12)H_{ax}$ , J = 3.2 Hz, J = 12.0 Hz, J = 12.0 Hz); 12.84 (s, 1 H, C(9)OH); 13.37 (s, 1 H, C(6)OH). <sup>13</sup>C NMR, δ: 18.4 (C(16)); 29.2 (C(17)); 33.5 (C(4)); 36.1 (C(15)); 61.6 (C(12));66.4 (C(13)); 72.9 (C(3)); 101.2 (C(2)); 108.8 (C(5a)); 110.1 (C(9a)); 121.2 (C(4a)); 132.7 (C(8)); 135.0 (C(7)); 156.2(C(10a)); 156.3 (C(6)); 157.3 (C(9)); 177.2 (C(5)); 183.2(C(10)). MS, m/z ( $I_{rel}$  (%)): 417 [M + 1]<sup>+</sup> (4.2), 416 [M]<sup>+</sup> (17.3), 415  $[M-1]^+$  (19.5), 414  $[M]^+$  (87.2), 413  $[M-1]^+$ (28.7), 412  $[M]^+$  (98.6), 411  $[M-1]^+$  (28.4), 330  $[M-1]^+$  $C_4H_6O_2$ ]<sup>+</sup> (6.5), 328 [M –  $C_4H_6O_2$ ]<sup>+</sup> (23.6), 326 [M –  $C_4H_6O_2$ ]<sup>+</sup> (33.9), 278  $[M - C_4H_6O_2 - C_4H_4]^+$  (8.3), 276  $[M - C_4H_6O_2 C_4H_4$ ]+ (9.4), 274 [M -  $C_4H_6O_2$  -  $C_4H_4$ ]+ (13.8), 69 (100).

(2S,3R,4R)-2,3-Ethylenedioxy-6,9-dihydroxy-7,8-dimethyl-2,4-trimethylene-3,4-dihydronaphtho[2,3-b]pyran-5,10(2H)**dione (3b).** Yield 63%,  $R_f$  0.53, m.p. 211–215 °C. Found (%): C, 64.30; H, 5.48. C<sub>20</sub>H<sub>20</sub>O<sub>7</sub>. Calculated (%): C, 64.51; H, 5.42. IR,  $v/cm^{-1}$ : 3450—2120 (OH); 1602 (C=O, C=C); 1580 (C=C); 1456; 1423; 1394; 1295; 1252; 1130. <sup>1</sup>H NMR, δ: 1.40—1.96 (m,  $5 H, C(16)H_{ax}, C(16)H_{eq}, C(15)H_{ax}, C(17)H_{ax}, C(17)H_{eq}); 2.18$ (ddt, 1 H, C(15) $H_{eq}$ , J = 1.3 Hz, J = 1.3 Hz, J = 5.2 Hz, J =13.8 Hz); 2.26 (s, 6 H, C(7)Me, C(8)Me); 3.53 (q, 1 H, C(4)H, J = 3.3 Hz); 3.63 (dd, 1 H, C(12)H<sub>eq</sub>, J = 2.7 Hz, J = 12.0 Hz); 3.75 (d, 1 H, C(3)H, J = 3.3 Hz); 3.76 (dd, 1 H, C(13)H<sub>eq</sub>, J =3.1 Hz, J = 12.0 Hz); 3.90 (ddd, 1 H, C(13)H<sub>ax</sub>, J = 2.7 Hz, J =12.0 Hz, J = 12.0 Hz); 4.24 (ddd, 1 H, C(12)H<sub>ax</sub>, J = 3.1 Hz, J = 12.0 Hz, J = 12.0 Hz; 13.06 (s, 1 H, C(9)OH); 13.42 (s, 1 H, C(6)OH). <sup>13</sup>C NMR, δ: 12.3 (C(18)); 12.6 (C(19)); 18.5 (C(16)); 29.4 (C(17)); 33.5 (C(4)); 36.3 (C(15)); 61.5 (C(12));66.4 (C(13)); 73.3 (C(3)); 100.3 (C(2)); 107.7 (C(5a)); 109.4 (C(9a)); 120.5 (C(4a)); 137.7 (C(8)); 140.4 (C(7)); 155.8(C(10a)); 161.0 (C(6)); 162.4 (C(9)); 175.9 (C(5)); 182.2(C(10)). MS, m/z ( $I_{rel}$  (%)): 373 [M + 1]<sup>+</sup> (23.2), 372 [M]<sup>+</sup> (100), 371  $[M - 1]^+$  (44.3), 286  $[M - C_4H_6O_2]^+$  (9.6), 234  $[M - C_4H_6O_2 - C_4H_4]^+$  (13.8)).

(2*S*,3*R*,4*R*)-2,3-Ethylenedioxy-6,9-dihydroxy-2,4-trimethylene-3,4-dihydronaphtho[2,3-b]pyran-5,10(2H)-dione (3c). Yield 44%,  $R_{\rm f}$  0.53, m.p. 203—205 °C. Found (%): C, 62.90; H, 4.59.

 $C_{18}H_{16}O_7$ . Calculated (%): C, 62.79; H, 4.68. IR,  $v/cm^{-1}$ : 3460-2190 (OH); 1608 (C=O, C=C); 1575 (C=C); 1504; 1457; 1286; 1249; 1147; 1123. <sup>1</sup>H NMR, δ: 1.40-1.96 (m, 5 H,  $C(16)H_{ax}$ ,  $C(16)H_{eq}$ ,  $C(15)H_{ax}$ ,  $C(17)H_{ax}$ ,  $C(17)H_{eq}$ ); 2.19 (ddt, 1 H, C(15)H<sub>eq</sub>, J = 1.3 Hz, J = 1.3 Hz, J = 5.2 Hz, J = 13.7 Hz); 3.54 (q, 1 H, C(4)H, J = 3.4 Hz); 3.66 (dd, 1 H, C(12) H<sub>eq</sub>, J = 2.7 Hz, J = 12.0 Hz; 3.76 (d, 1 H, C(3)H, J = 3.4 Hz); 3.77 (dd, 1 H, C(13) $H_{eq}$ , J = 2.7 Hz, J = 12.0 Hz); 3.91 (ddd, 1 H,  $C(13)H_{ax}$ , J = 2.7 Hz, J = 12.0 Hz, J = 12.0 Hz); 4.23 (ddd, 1 H,  $C(12)H_{ax}$ , J = 2.7 Hz, J = 12.0 Hz, J = 12.0 Hz); 7.18 (d, 1 H, C(8)H, J = 10.6 Hz); 7.26 (d, 1 H, C(7)H, J = 10.6 Hz); 12.29 (s, 1 H, C(9)OH); 12.72 (s, 1 H, C(6)OH). <sup>13</sup>C NMR, δ: 18.5 (C(16)); 29.3 (C(17)); 33.4 (C(4)); 36.2 (C(15)); 61.6 (C(12)); 66.4 (C(13)); 73.1 (C(3)); 100.8 (C(2)); 110.9 (C(5a)); 111.7 (C(9a)); 121.9 (C(4a)); 128.3 (C(8)); 130.5 (C(7)); 156.4 (C(10a)); 157.1 (C(6)); 158.4 (C(9)); 181.1 (C(5)); 186.7(C(10)). MS, m/z ( $I_{rel}$  (%)): 345 [M + 1]<sup>+</sup> (3.4), 344 [M]<sup>+</sup> (11.5), 258  $[M - C_4H_6O_2]^+$  (5.6), 206  $[M - C_4H_6O_2 - C_4H_4]^+$ (19.8), 59 (100).

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